

### **UNCONVENTIONAL RESERVOIR ENGINEERING PROJECT**

COLORADO SCHOOL OF MINES

**Research Summary** 

### Modeling Hydrocarbon Filtration in Nanoporous Media

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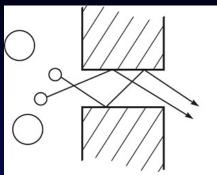


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### **Background - Membrane Transport Theory**

### Molecular transport through membranes:

- by a flow through permanent pores
- by the solution-diffusion mechanism



Microporous membranes separate by molecular filtration

Dense solution-diffusion membranes separate because of differences in the solubility and mobility of permeants dissolved in the membrane material

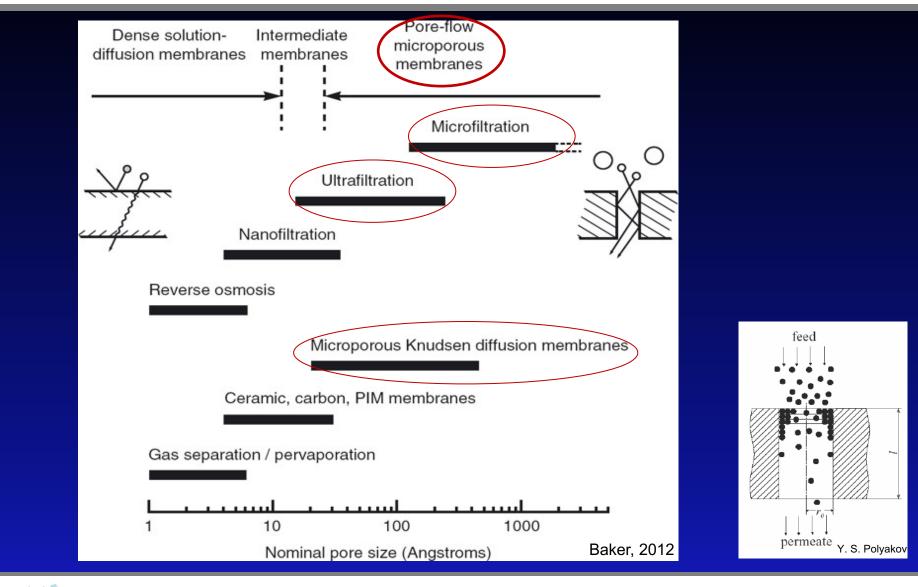
**Pore-flow model:** Transported by pressure-driven convective flow though nanoporous media.

<u>Solution-diffusion model</u>: Dissolve in the membrane material & diffuse though the membrane down a concentration gradient.

Baker, 2012



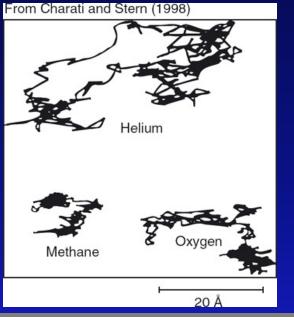
### Background - Membrane Transport Theory (cont`d)



### **Solution-Diffusion Model**

- Applies to reverse osmosis, pervaporation, and gas permeation in polymer films
- Provides equations correctly link concentration driving forces, pressure with flux, and selectivity
- Composition, pressure, and temperature of the fluids: determine the concentration of the diffusing species

The movements of 3 molecules by computational techniques:



Helium molecule:

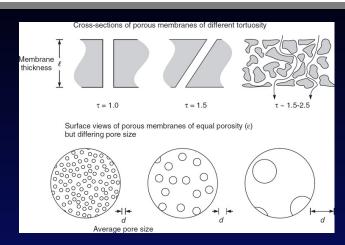
- Smaller
- moves more frequently
- makes larger jumps



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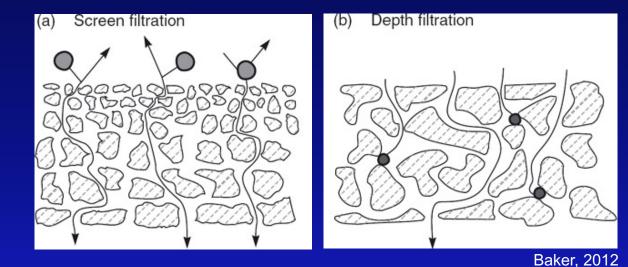
### **Pore-Flow Model**

- Characterizing the complexity of microporous membranes – imperfect
- Mostly used parameters: tortuosity (τ), membrane porosity (ε) & average pore diameter (d)
- The most important: pore diameter

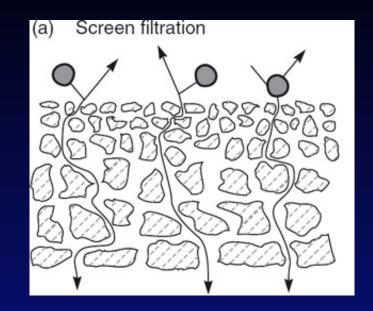


#### Separation of particulates:

- at the membrane surface – screen filtration mechanism
- in the interior of the membrane – depth filtration mechanism



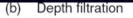
### **Pore-Flow Model – Screen filtration**

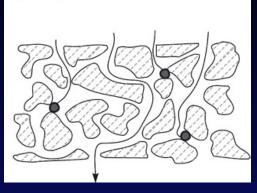


- The mechanism is relatively easily described.
- Surface pores are smaller than the particles to be removed.
- Small particles pass through the surface pores & not captured in the interior of the membrane.
- Ultrafiltration membranes



## **Pore-Flow Model – Depth filtration**

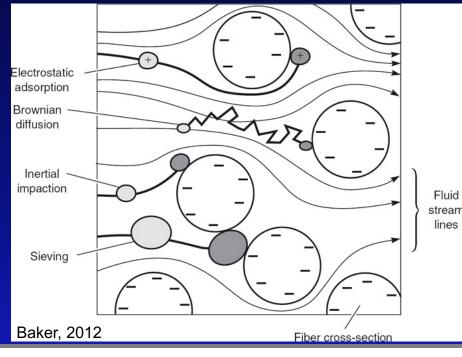




- The mechanism is more complex
- Captures the particles to be removed in the interior of  $\bullet$ the membrane
- Microfiltration membranes ullet
- 4 mechanisms contribute to capture particle  $\bullet$

### Sieving:

- ✓ captures particles at constrictions Inertial impaction:
- ✓ capture particles by adsorption
- $\checkmark$  most efficient for larger particles Brownian diffusion:
- ✓ capture particles by surface adsorption
- ✓ most efficient for smaller particles *Electrostatic adsorption:*
- ✓ capture of charged particles





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Fluid

lines

### **Pore-Flow Model - Motivation**

- Transport modeling through microporous membranes in ultrafiltration & microfiltration is less developed
- Permeation

affected by several effects – hard to compute function of membrane structure and composition

Current theories cannot fully model the permeation properties of the membranes



Empirical Darcy or Fick's Law relate diffusive flux to the gradient of a process variable (pressure, concentration, temperature, etc.)

Thermodynamically, diffusion occurs so as to minimize the free energy. Therefore, a more general expression for diffusive flux is given in terms of the chemical potential gradient.

$$J_i = -L_i \frac{\partial \mu_i}{\partial x}$$

If the driving forces are concentration and pressure

$$d\mu_i = RTd\ln(\gamma_i n_i) + v_i(p - p_i^0)$$
  
$$\mu_i = \mu_i^0 RTd\ln(\gamma_i n_i) + v_i(p - p_{i_{sat}})$$

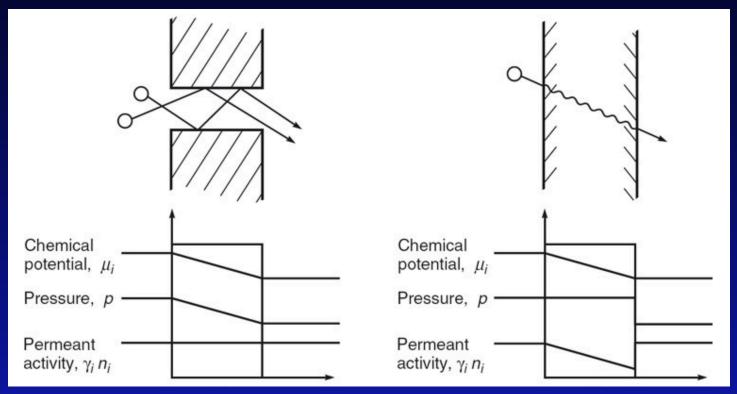
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### **Transport in Nanoporous Media**

#### Assumptions involved in filtration modeling

**Pore-Flow Model** 

Solution-Diffusion Model



 $\mu_i = \mu_i^0 RTd \ln(\gamma_i n_i) + v_i (p - p_{i_{sat}})$ 



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## **Literature Review**

#### Ruthven et al., 1973:

- Reported experimental results of sorption and diffusion of light HCs
- Diffusion activation energy proves a good correlation with critical diameter

### Shelekhin et al., 1995:

- Proposed a mathematical model to define diffusion, permeation and separation of gases in microporous molecular-sieve membranes
- Showed that the diffusion activation energy increases with increasing kinetic diameter of the diffusant

#### Polyakov et al., 2002:

- Studied a mathematical model of a depth filter
- Used the solution to analyze the depth filter performance as a function of the size of filtration velocity, colloidal particles, specific filter surface, and filter depth

#### <u>Smit, B., 2007</u>:

- Demonstrated how molecular simulation can be useful in order to understand behavior of molecules inside the pores of zeolites
- Showed used rare event simulation techniques with practical use and importance



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## **Literature Review**

### Polyakov, Y. S., 2008:

- Developed a macroscopic depth filtration theory for the standard blocking of the pores of 'Ultrafiltration'/'Microfiltration' membranes
- The theory guesses the membrane permeation rate and selectivity & shows the profile of the number of deposited particles
- The proposed formulations were confirmed with the experimental data

#### Sampath et al., 2014:

- Studied practicability of using different proposed models for filtration of a satisfying product at constant pressure
- Observed: smaller pore filters experience fouling through complete pore blocking larger pore filters experience fouling through intermediate pore blocking
- Recommended: "The proposed approach can be used for more accurate sizing of microfilters and depth filters"

#### Kuhn et al., 2017:

- Introduced a new model-based method to improve depth filters design
- Confirmed the numerical optimal control algorithm using a newly developed analytical solution of local filtration performance



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### **Questions?**

# Thank you

